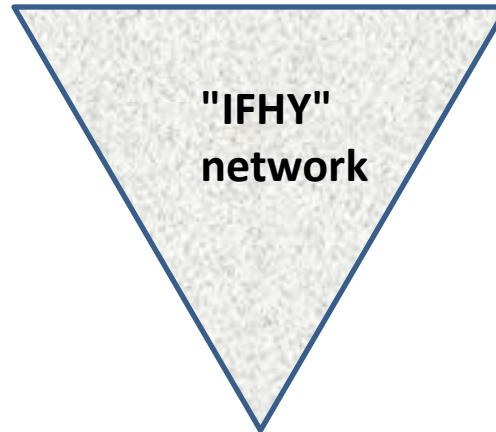


# Hydrogen in France:



**& Universities**



**Industries**



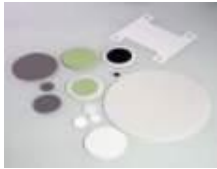
Prof. Olivier Joubert  
[olivier.joubert@cns-imn.fr](mailto:olivier.joubert@cns-imn.fr)  
[www.gdr-hypac.cns.fr](http://www.gdr-hypac.cns.fr)

# Hydrogen research in France:

**MATERIALS**



**PROCESSES**



**COMPONENTS**



**SYSTEM INTEGRATION**



**DEMONSTRATION**



**FIRST PROTOTYPES  
MANUFACTURING**

Including  
Modeling, Characterization and Diagnostic



# HySPaC

Head : Prof. Olivier Joubert  
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[www.gdr-hyspac.cnrs.fr](http://www.gdr-hyspac.cnrs.fr)

French academic research grouping on Hydrogen

**HySPaC:** « Hydrogen. Systems. Fuel Cells » **A Structuring « academic » research operation (started in 2001 !)**

**Objectives:**

- Promote scientific workshops & networking
- Ensure a permanent updated state-of-the-art
- Establish realistic scientific objectives for French research centers
- Build consortia for applying to national or international programs

- 250 researchers
- 4 topics (Low and High temperature FC and E, Hydrogen Production (not E) and Storage, FC and E Systems)
- Scientific shared production (/year) : 50 papers, 25 PhD, 5 patents,



**Members**

- 77 CNRS teams
- 10 CEA teams
- industries

**Competences**

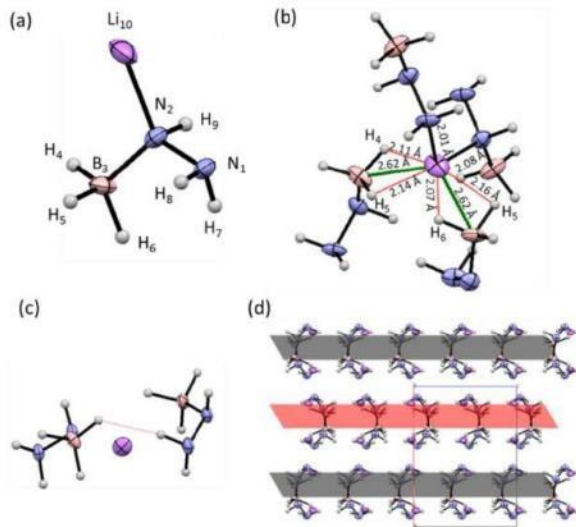
- Materials,
- shaping,
- caraterisations : Electrochemistry, Mecanical, thermal, performance
- modelisation...

## Scientific & Technical Challenges

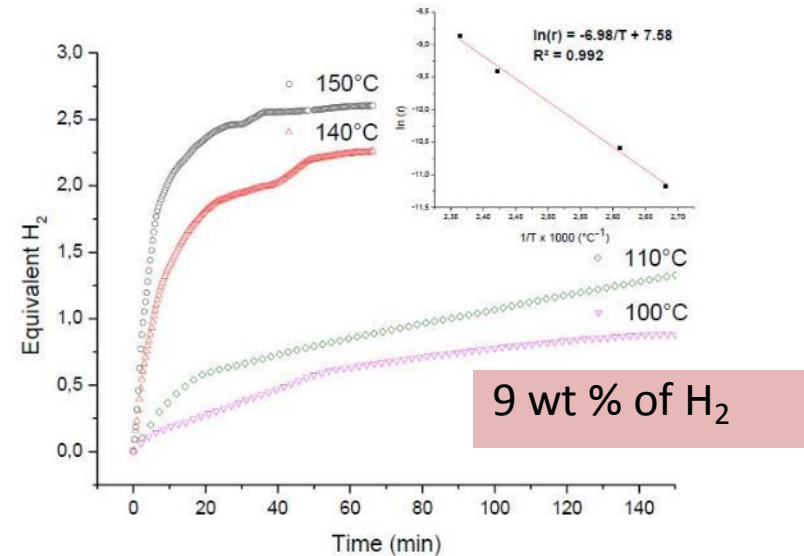
- Hydrogen production methods with low CO<sub>2</sub> fingerprint
- Hydrogen purification by selective membranes
- Materials for solid-state hydrogen storage
- Development of storage tanks
- Coupling of storage tanks to fuel cells systems

[fermin.cuevas@icmpe.cnrs.fr](mailto:fermin.cuevas@icmpe.cnrs.fr)

## Light B–H and B–N–H compounds for energy



**Figure 4.** Crystal structure of  $\beta$ -LiHB with detailed view of: (a) the asymmetric unit with labels; (b) the coordination of  $\text{Li}^+$ ; (c) the intermolecular  $\text{H}^{\delta+}\cdots\text{H}^{\delta-}$  dihydrogen bond according to the Klooster et al.'s definition<sup>20</sup> ( $\text{H}_4\cdots\text{H}_7 = 2.25 \text{ \AA}$ ,  $\text{B}_3\text{-H}_4\cdots\text{H}_7 = 106.8^\circ$  and  $\text{N}_1\text{-H}_7\cdots\text{H}_4 = 171.2^\circ$ ); and (d) the representation of the parallel planes on which the  $\text{H}^{\delta+}\cdots\text{H}^{\delta-}$  network extend.



**Figure 6.** Volumetric measurements under isothermal treatment at 100, 110, 140 and 150°C of  $\beta$ -LiHB. The inset figure is the Arrhenius plot for the determination of the apparent activation energy from the data of the volumetric measurements.

*Our focus:* B–H and B–N–H materials, from chemical hydrogen storage materials to liquid-state fuel of fuel cell, via energetic materials...

# HySPaC Hydrogen storage: energy management

Fuel Cells, H<sub>2</sub>, Renewable Energy Management

## Example of a R&D program

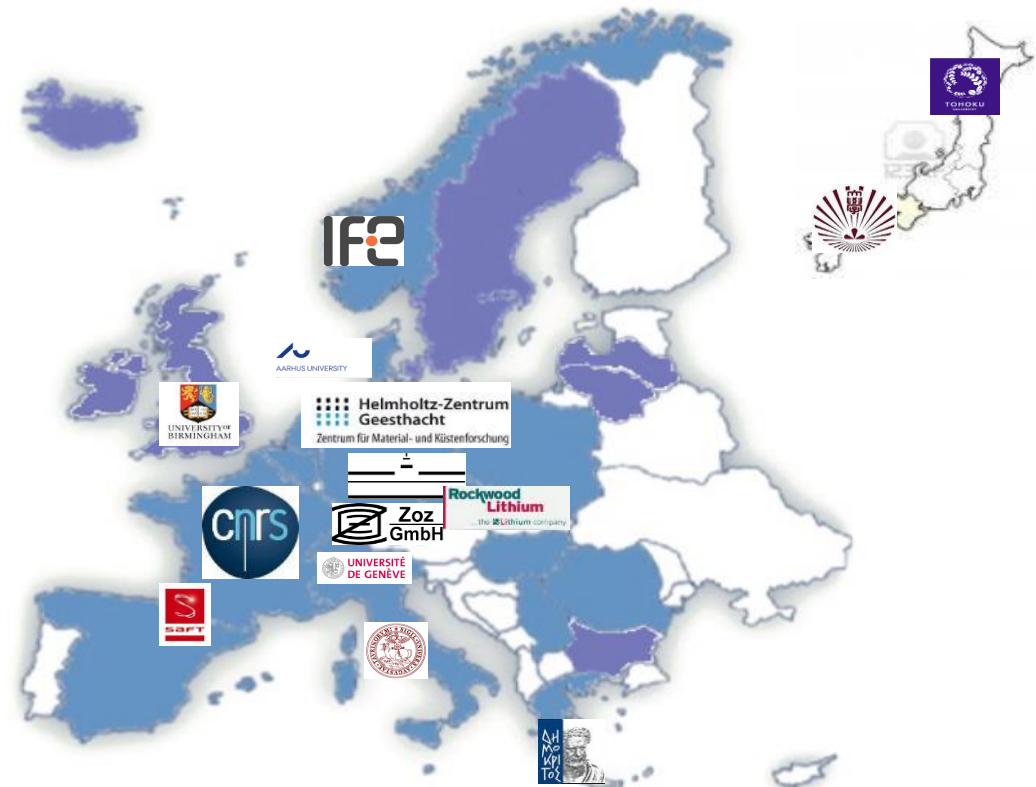


the ECOSTORE project (2013-2017)

*ITN Marie Curie Action: 4 M€*

Novel Complex Metal Hydrides  
for Efficient and Compact  
Storage of Renewable Energy a  
Hydrogen and Electricity

13 European laboratories  
and industries  
2 Japanese labs



[latroche@icmpe.cnrs.fr](mailto:latroche@icmpe.cnrs.fr)

## Stand-alone relay station

H<sub>2</sub> storage by pressurized gas



type-IV tanks  
5 kgH<sub>2</sub>, 30 bar

Modular all-in one energy  
station ensuring continuous  
seasonal autonomy

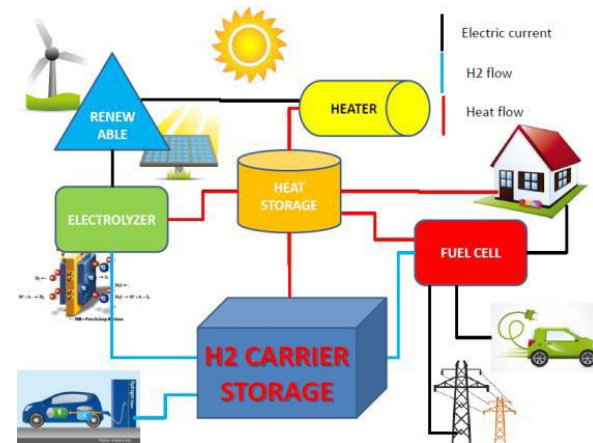
Collaborative project RHYTA (2015-2018)



[david.chapelle@univ-fcomte.fr](mailto:david.chapelle@univ-fcomte.fr)

## Stationary storage

H<sub>2</sub> storage in metal hydrides



TiFeH<sub>2</sub> and PCM materials for the  
storage of renewable energy  
50 kgH<sub>2</sub>, 2 bar

HyCARE project (2019-2021)



[latroche@icmpe.cnrs.fr](mailto:latroche@icmpe.cnrs.fr)



# HySPaC Fuel Cells and electrolysers

## Scientific & Technical Challenges

Expensive membranes

Need of Pt catalyst

Complex Reforming  
at low T (no CO)

Water management,  
Hydration of gases /  
Limited cogeneration

Low cost fabrication processes (metallic  
materials)

Easier thermal cycling

Simple reforming

**BUT Need of materials for Electrolytes  
and  
Electrodes**

Expensive interconnects

(ceramics or alloys)

Difficulty of thermal cycling

Too long starting time

**Degradation of electrodes  
at high temperature**

Sealing problems. Lifetime

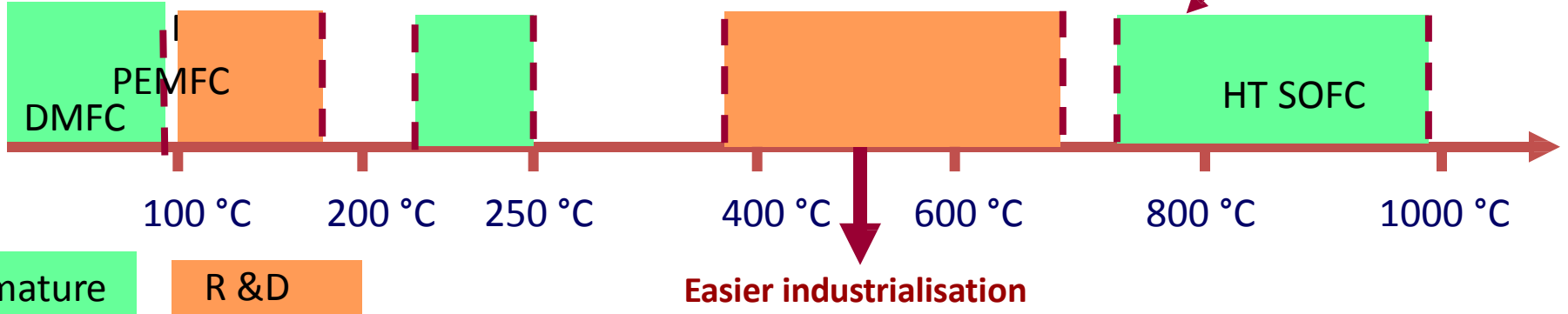
Polymers

Stainless steel

Ceramics

Tolerance to CO

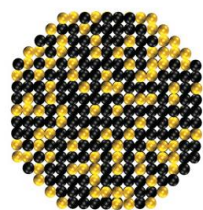
Degradation



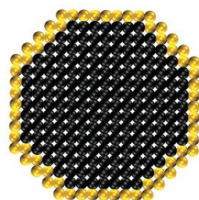


Development of synthesis methods for active, selective and stable catalysts for the **ORR** and the oxidation of small molecules with energetic interest

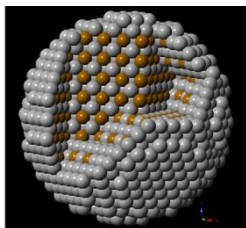
## PGM-based Catalysts



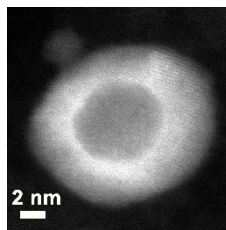
alloys



Core-shell

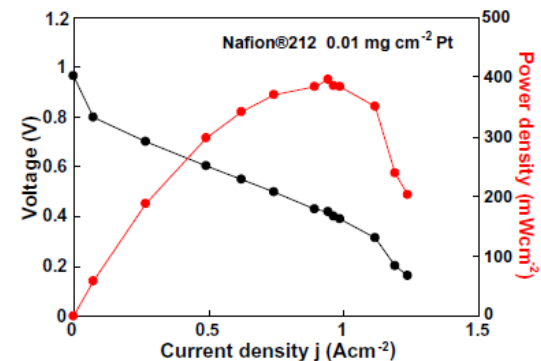
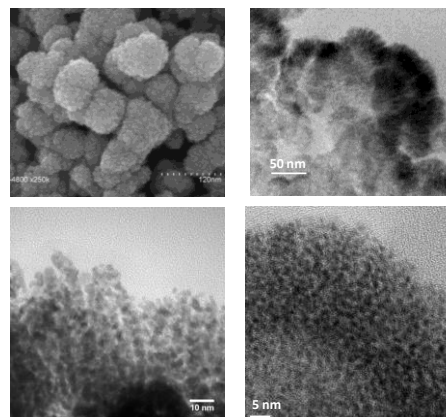


intermetallic



"Hollow" Pt NPs

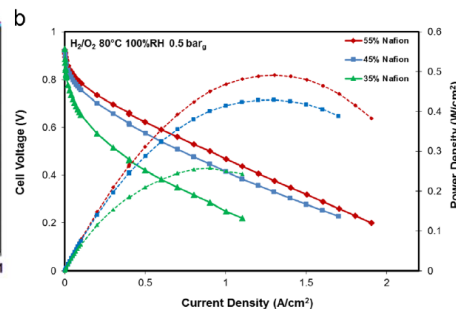
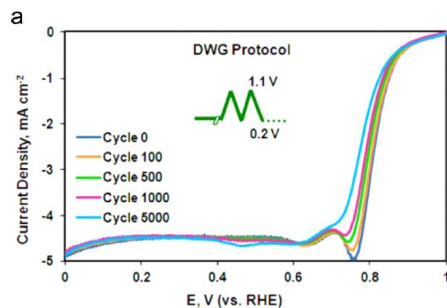
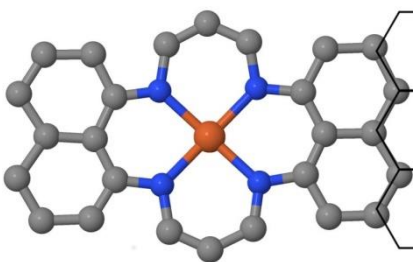
## New electrode architecture



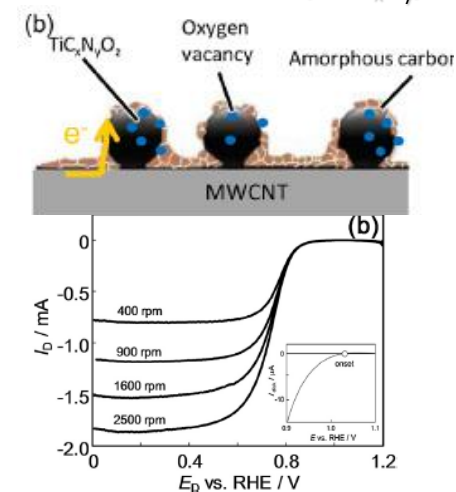
0.02 mg<sub>Pt</sub> cm<sup>-2</sup>; 0.4 W cm<sup>-2</sup> => 20 kW<sub>e</sub> /g<sub>Pt</sub>

M. Cavarroc, et al. *Electrochem. Comm.* 11 (2009) 859–861 ;  
P. Brault et al. *ChemSusChem* 6 (2013) 1168-1171.

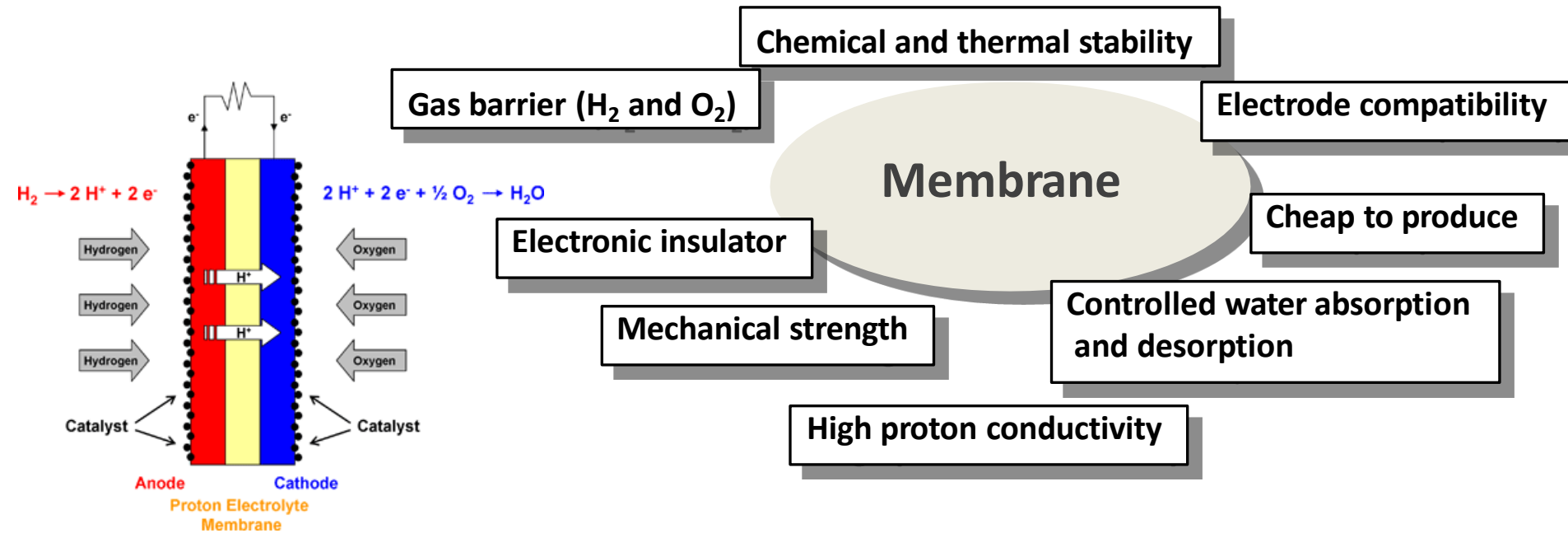
## No-PGM based catalysts



## Titanium carbonitride (TiCN<sub>x</sub>O<sub>y</sub>)



## Fuel Cells, H<sub>2</sub>, Renewable Energy Management

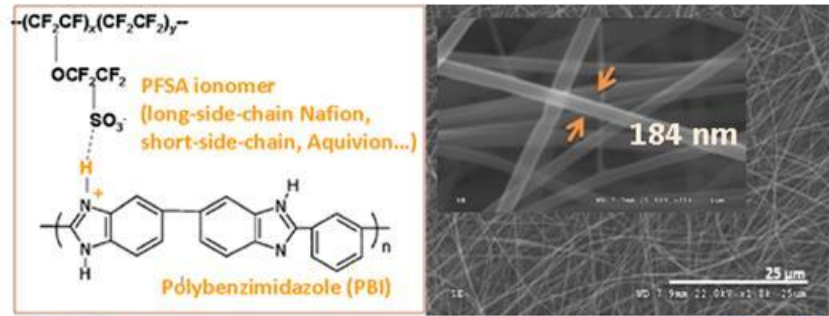


- ➔ Develop alternative membranes to PFSA membrane: low cost and with good thermo-mechanical strength, conductivity in a large window of T ( 0 < T > 120°C) and low RH
- ➔ Increase the operation temperature of PEMFC at more than 120°C at low RH or in anhydrous conditions
- ➔ Study/Enhance the durability of MEA

# HySPaC PEM (Membranes)

## Robust Membrane Development

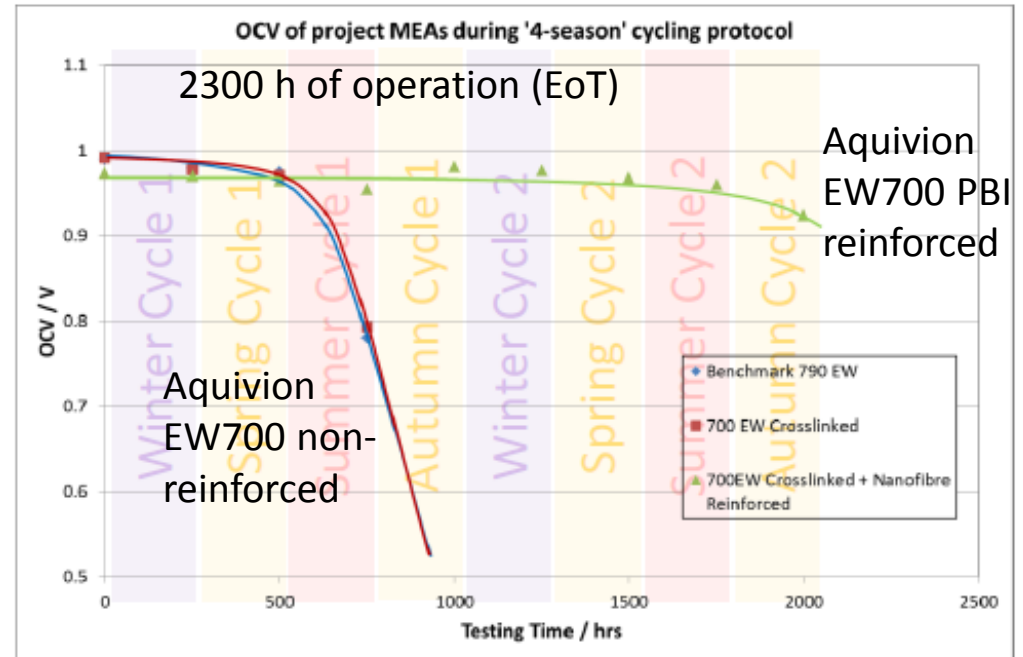
ICGM | Département de Chimie du Solide et de la Matière Divisée | AIME  
Institut Charles Gerhardt Montpellier



- Composite membranes PFSA/nanofibres (with PBI)
- Separation of the two functions: structural (reinforce) and transport (ionomer)
- "ionic Reticulation" good interface between materials
- Stack "arc-en-ciel" 9 cells, 50 cm<sup>2</sup>, 80 °C, 30% RH.



Scaling up of reinforced membrane



FCH JU MAESTRO and VOLUMETRIQ Projects

## scientific and technological Hurdles

### MEA:

#### New materials for SOFC and SOEC

- **Electrolytes** : composites, proton conductors (stability vs conductivity), modélisation (DFT,...) vs performance
- **Electrodes** : MIEC (particularly ionic conductivity  $10^{-2}$ S/cm), stability (vs CO<sub>2</sub>), decreasing polarization resistance (microstructure, catalyst)
- **interconnectors** (more and more important / metal supported cell) : solving the problem of oxidation during the design of the cell, using precursors to synthesize interconnects, how to limit the Cr diffusion (development of protecting layer).
- **sealings** : glass type new materials to improve stability, self-healing

### Half and Complete cells:

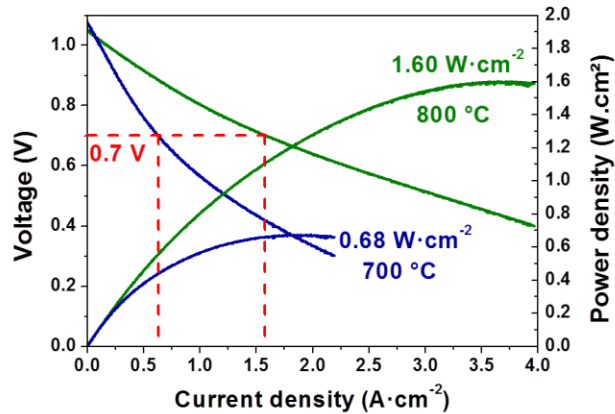
- Using **industrials shaping methods**
- **New diagnostic tools** (electrochemical tools)
- Cost reducing: reducing the number of heating steps

### Tests, durability (vs impurities):

- improving the performance by adding diffusion barrier or by modifying the composition
- **modelisation** (degradation mechanism)

### Innovative oxygen and impregnated electrodes

1) Ni-YSZ//YSZ//GDC//Pr<sub>4</sub>Ni<sub>3</sub>O<sub>10+δ</sub>  
Screen-printed electrodes



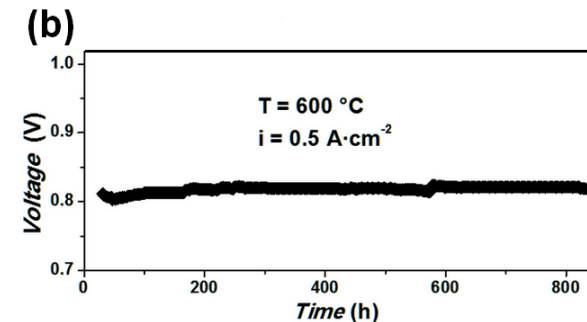
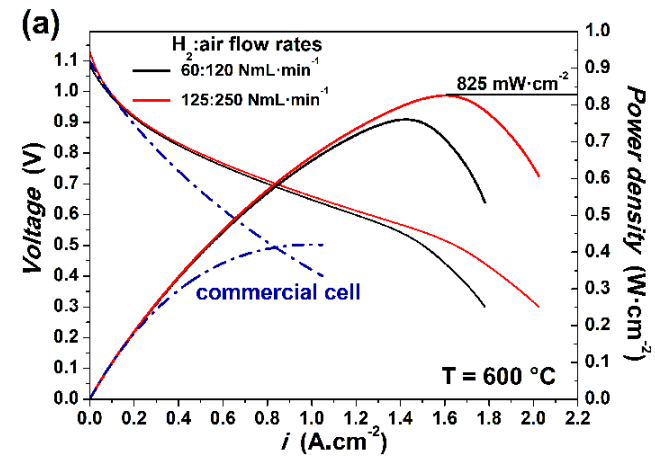
Power density  
At 800 °C: 1.6 W.cm<sup>-2</sup>

3) Innovative cells

1) Shaping of a porous/dense/porous support using the same electrolyte

2) Infiltration of solutions => preparation of electrodes at moderate T°

2) Porous CGO impregnated by Pr<sub>6</sub>O<sub>11</sub>  
Benchmark with commercial cell (600°C)



[jean-marc.bassat@icmcb.cnrs.fr](mailto:jean-marc.bassat@icmcb.cnrs.fr)

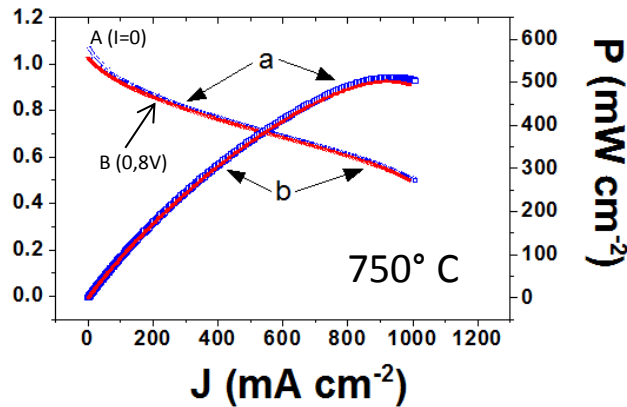
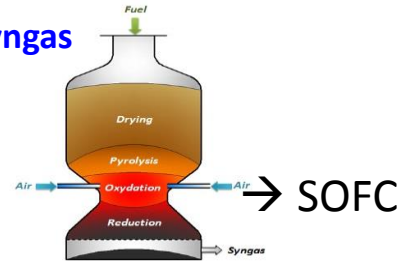
## TEST: Electrochemical diagnosis

VALORPAC and Roxsolidcell \* Projects



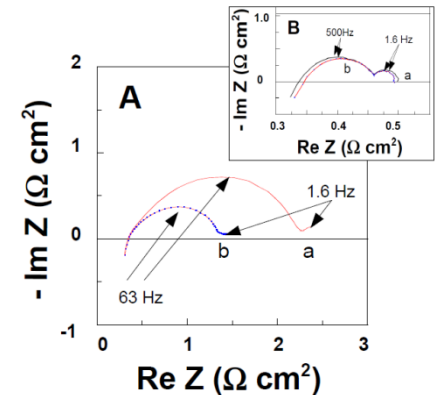
Using SOFC with syngas

→ Gazifier



(a) H<sub>2</sub> 20% - CO 20% - CO<sub>2</sub> 10% - N<sub>2</sub> 50% : 482 mW cm<sup>-2</sup>

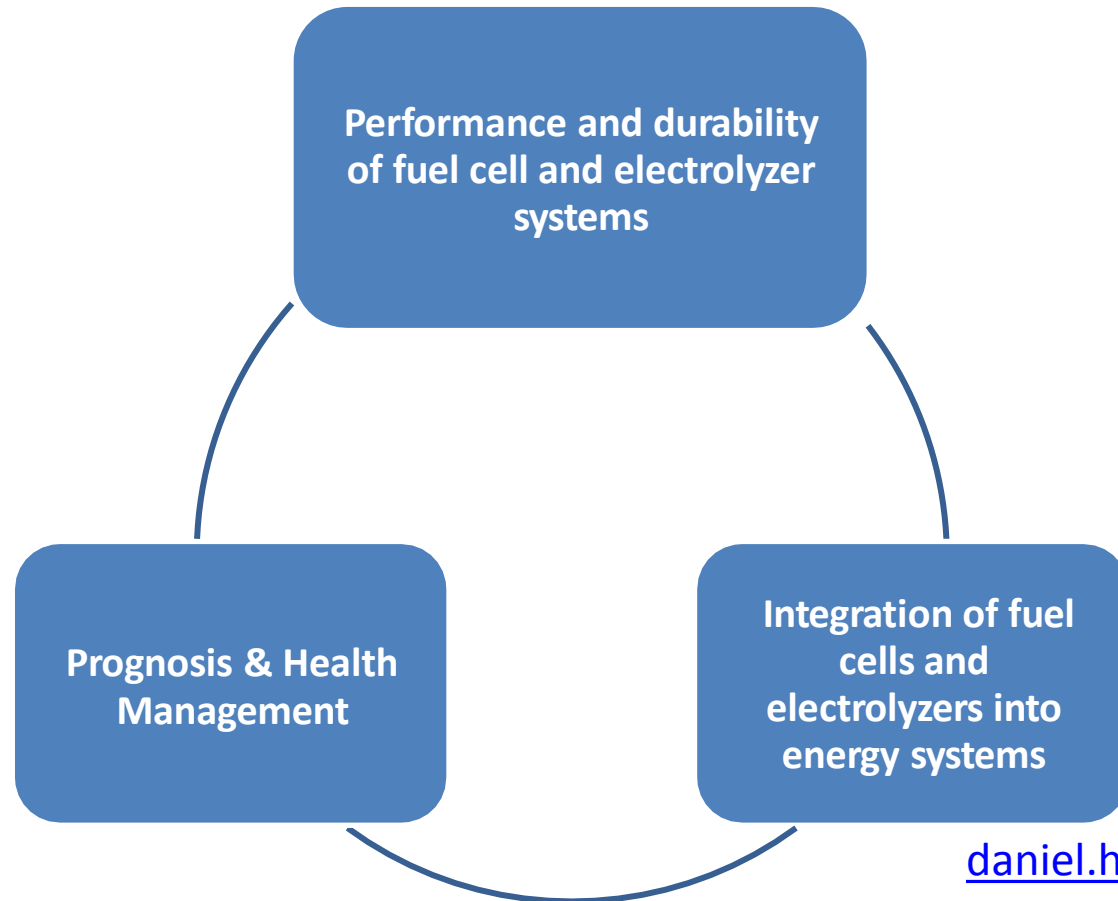
(b) H<sub>2</sub> 17% - CH<sub>4</sub> 3% - CO 20% - CO<sub>2</sub> 10% - N<sub>2</sub> 50% : 502 mW cm<sup>-2</sup>



Impedance diagrams →  
Electrodes :  
80% of ASR at OCV, 30% at 0.8V

"Effects of carbon monoxide, carbon dioxide, and methane on Nickel/Yttria-stabilized Zirconia-based Solid Oxide Fuel Cells performance for direct coupling with a gasifier" M. Lebreton. B. Delanoue . E. Baron F. Ricoul A. Kerihuel A. Subrenat, O. Joubert, A. Le Gal La Salle, *International Journal of Hydrogen Energy*, 40 (32), 10231-10241 (2015)

Multi-flow energy systems based on fuel cells and electrolyzers



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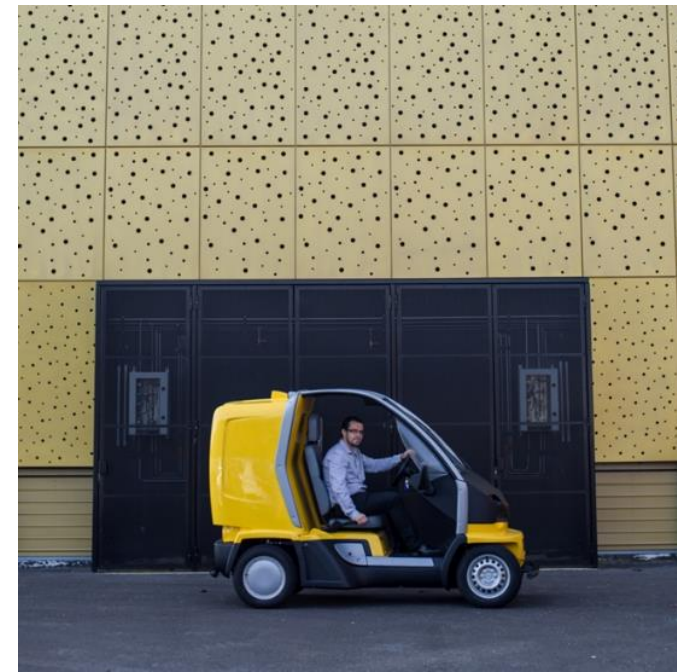
## Mobypost EU project – Main project objectives

- Taking care of postmen requirements
- Design an optimized hybrid FC powertrain
- Energy flow supervision
- Coupling with renewables (PV panels)
- Hydrogen production & storage on-site
- 1st French FCV fleet (10 vehicles, 2 H2/PV stations) testing in constrained environment (temperature, power demand)



## Key numbers

- **2** demonstration territories in B-FC region
- **2** years experimental trial
- **8** European partners
- **10** FC vehicles
- **920** MM work
- **1682** postal routes covered
- **2017** (demonstration ended in...)





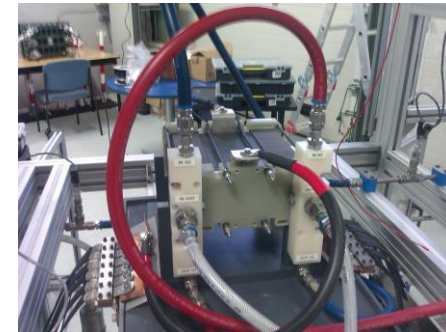
## FC LAB (Belfort)

- 60 permanent staff
- Facilities area > 1000 m<sup>2</sup>
- **Land transport** applications
- Stationary applications
- Test up to 100kW



## (Toulouse) **Laplace**

- 13 permanent staff
- Facilities area: 250m<sup>2</sup>
- **Aircraft** applications
- Stationary applications
- Test up to 10kW



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**Thank you for your attention**

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